

Electrostatics Notes

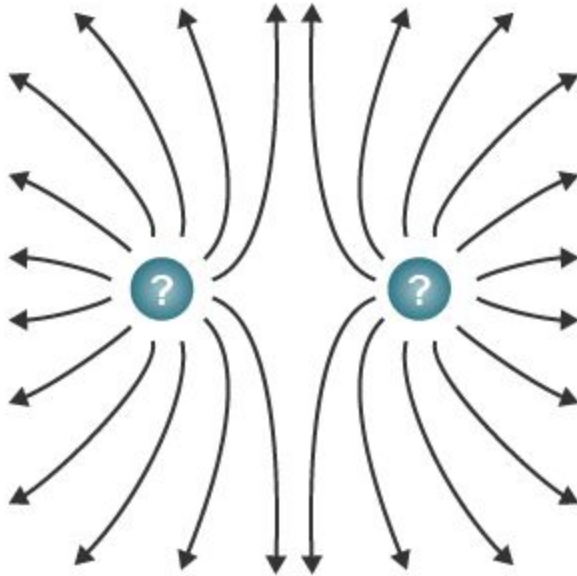
- **Electrostatics** - study of charged particles at rest
- **Ion** - atom or molecule with a net charge
 - Due to loss or gain of electrons
 - Can be passed from solar winds or Earth's core
- **Question**: An oxygen atom picks up two additional, free-floating electrons. Is the charge of the newly formed oxygen ion positive, negative, or neutral?
 - Negative, electrons have negative charges so the more of them you have the more negatively charged the ion will be.
- **Electric field** - the area around a charged object that can exert a force on other charged objects
- **Electric force** - force between two charged objects

○ Equation:

$$F = qE$$

- F = electric force
 - Unit: N
- Q = electric charge
 - Unit: coulomb (C)
- E = electric field
 - Unit: N/C
- **Question**: A charge of 4.5×10^{-5} C is placed in an electric field with a strength of 2.0×10^4 N/C. What is the electric force acting on the charge?
 - Given: $q = 4.5 \times 10^{-5}$ C
 $E = 2.0 \times 10^4$ N/C
 - Unknown: $F = ?$
 - Equation: $F = qE$
 - Substitute: $F = (4.5 \times 10^{-5} \text{ C})(2.0 \times 10^4 \text{ N/C})$
 - Solve: $F = 0.9$ N
- **Field lines** - lines in a diagram that show the direction of flow of the electric field between two charged particles
 - Point away from positive
 - Point toward negative
 - When two charges are near each other:
 - Like charges bend away (repel)
 - Opposite charges combine (attract)

- **Question:** Based on the field lines, the electric charges indicated by the question marks are _____.



- The same. Like charges bend away from (repel) each other when they are close.
- Electrically charged particles or ions can behave differently when they enter a magnetic field.
- **Electric potential energy** - potential energy an electric charge has due to its location in an electric field

$$U = Fd$$

- **Equation:**
 - U = electric potential energy
 - Unit: Joule (J)
 - F = electric force
 - Unit: N
 - d = distance between particle and source of electric field
 - Unit: m
- **Question:** A charge of 4.5×10^{-5} C is placed in an electric field with a strength of 2.0×10^4 N/C. If the charge is 0.030 m from the source of the electric field, what is the electric potential energy of the charge?
 - Given: $q = 4.5 \times 10^{-5}$ C
 $E = 2.0 \times 10^4$ N/C
 $d = 0.030$ m
 - Unknown: $U = ?$
 - Equation: $U = qEd$

- Substitute: $U = (4.5 \times 10^{-5} \text{ C})(2.0 \times 10^4 \text{ N/C})(0.030 \text{ m})$
- Solve: $U = 0.027 \text{ J}$
- Electric potential - electrical potential energy of a charged particle divided by its charge



$$V = \frac{U}{q}$$

- Equation:
 - V = electric potential
 - Unit: Volt (V)
 - U = electric potential energy
 - Unit: J
 - q = electric charge
 - Unit: C
- Question: What is the electric potential of a $4.5 \times 10^{-5} \text{ C}$ charge that has an electric potential energy of 0.027 J ?
 - Given: $q = 4.5 \times 10^{-5} \text{ C}$
 $U = 0.027 \text{ J}$
 - Unknown: $V = ?$
 - Equation: $V = U/q$
 - Substitute: $V = (0.027 \text{ J})/(4.5 \times 10^{-5} \text{ C})$
 - Solve: $V = 600 \text{ V}$
- Conductor - any material that allows electricity or thermal energy to easily move through it
 - Ex. Metals, water, ionic solutions

Periodic Table of Elements

Click an element's box for more info.

Legend:

- Non-Metal
- Semi-Metal
- Alkali Metal
- Alkaline Earth Metal
- Transition Metal
- Post-Transition Metal
- Halogen
- Noble Gas
- Lanthanide
- Actinide

| 1 | 2 | 3 | 4 | 5 | 6 | 7 | 8 | 9 | 10 | 11 | 12 | 13 | 14 | 15 | 16 | 17 | 18 |
|-------|-------|-------|-------|-------|-------|-------|-------|-------|-------|-------|-------|-------|--------|--------|--------|--------|-------|
| 1A | 2A | 3B | 4B | 5B | 6B | 7B | 8B | 9B | 10B | 11B | 12B | 3A | 4A | 5A | 6A | 7A | 8A |
| 1 H | 2 He | | | | | | | | | | | 3 B | 4 C | 5 N | 6 O | 7 F | 8 Ne |
| 3 Li | 4 Be | | | | | | | | | | | 9 Al | 10 Si | 11 P | 12 S | 13 Cl | 14 Ar |
| 11 Na | 12 Mg | | | | | | | | | | | 13 Ga | 14 Ge | 15 As | 16 Se | 17 Br | 18 Kr |
| 19 K | 20 Ca | 21 Sc | 22 Ti | 23 V | 24 Cr | 25 Mn | 26 Fe | 27 Co | 28 Ni | 29 Cu | 30 Zn | 31 Ga | 32 Ge | 33 As | 34 Se | 35 Br | 36 Kr |
| 37 Rb | 38 Sr | 39 Y | 40 Zr | 41 Nb | 42 Mo | 43 Tc | 44 Ru | 45 Rh | 46 Pd | 47 Ag | 48 Cd | 49 In | 50 Sn | 51 Sb | 52 Te | 53 I | 54 Xe |
| 55 Cs | 56 Ba | 57 La | 58 Ce | 59 Pr | 60 Nd | 61 Pm | 62 Sm | 63 Eu | 64 Gd | 65 Tb | 66 Dy | 67 Ho | 68 Er | 69 Tm | 70 Yb | 71 Lu | 72 Hf |
| 87 Fr | 88 Ra | 89 Ac | 90 Th | 91 Pa | 92 U | 93 Np | 94 Pu | 95 Am | 96 Cm | 97 Bk | 98 Cf | 99 Es | 100 Fm | 101 Md | 102 No | 103 Lr | |

Just about everything to the left of this squiggly line is a metal

- Insulator - material that restricts the flow of electricity or thermal energy
 - Ex. rubber, glass, wood
- Question: Classify each substance as either a conductor or insulator.
 - A sample of mercury:
 - Conductor, it is a metal since it is to the left of the squiggly line
 - A piece of glass:
 - Insulator, that was one of the examples
 - A rubber hose:
 - Insulator, that was one of the examples
 - A negatively ionized lithium paste:
 - Conductor, it is a metal since it is to the left of the squiggly line
- You can charge objects by friction, a.k.a. rubbing things together
 - Ex. rubbing a glass rod with silk causes electrons to go from rod to silk fabric
 - Now the silk is negative and the rod is positive
- **Conduction** - electrons are transferred from one object to another by direct contact
- **Induction** - electric charges are transferred with nothing touching